

# PATENT SPECIFICATION

(11) 1374337



- 2 (21) Application No. 11830/72 (22) Filed 14 March 1972  
 3 (44) Complete Specification published 20 Nov. 1974  
 4 (51) International Classification C07D 7/40 A61K 27/00//C07C 103/22  
 5 C07D 99/04

1 (52) Index at acceptance

C2C 1341 1494 1532 1562 1582 1626 1673 200 213 215  
 220 226 227 22Y 246 247 250 251 252 253 255  
 256 25Y 280 282 28X 305 30Y 311 313 31Y 323  
 32Y 337 338 342 34Y 351 352 360 363 364 36Y  
 386 405 40Y 43X 583 584 593 620 623 624 625  
 62X 650 652 662 672 694 699 761 767 790 79Y  
 KH KQ LF LK NM TU ZD

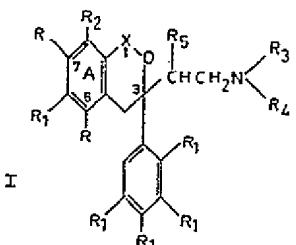
(72) Inventors WILLIAM J. HOULIHAN and  
 JEFFREY NADELSON

(54) AMINOALKYL-ISOCHEROMANS AND ISOCOUMARINS

(71) We, SANDOZ LTD., of 35  
 Lichtstrasse, 4000 Basle, Switzerland, a Swiss  
 Body Corporate, do hereby declare the invention,  
 for which we pray that a patent may be  
 granted to us, and the method by which it is  
 to be performed, to be particularly described  
 in and by the following statement:—

The present invention relates to novel  
 tertiary aminoethyl isochromans and iso-  
 coumarins.

The invention provides compounds of  
 formula I,



in which each  
 15 R independently signifies hydrogen, trifluoromethyl or alkyl or alkoxy of 1 to 5 carbon atoms, each  
 R<sub>1</sub> independently signifies hydrogen, fluorine, chlorine, trifluoromethyl or alkyl or alkoxy of 1 to 5 carbon atoms, or two R<sub>1</sub>'s on adjacent carbon atoms together signify methylenedioxy,  
 20 R<sub>2</sub> signifies hydrogen, trifluoromethyl, alkoxy of 1 to 5 carbon atoms, fluorine or chlorine,  
 R<sub>3</sub> and R<sub>4</sub> independently signify alkyl of 1 to 5 carbon atoms, alkenyl of 2 to 5 carbon atoms or benzyl, or  
 25 R<sub>3</sub> and R<sub>4</sub> together signify a -(CH<sub>2</sub>)— chain of 4 to 7 carbon atoms or  
 —(CH<sub>2</sub>)<sub>2</sub>—Z—(CH<sub>2</sub>)<sub>2</sub>—

in which Z signifies oxygen or sulphur or nitrogen substituted by alkyl of 1 to 5 carbon atoms

R<sub>5</sub> signifies hydrogen or straight chain alkyl of 1 to 5 carbon atoms and

X signifies —CH<sub>2</sub>— or —CO—, with the provisos that

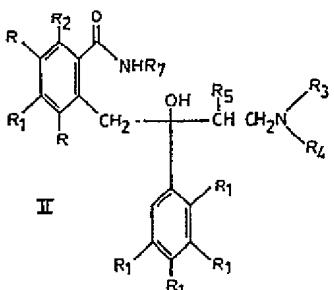
i) no more than three of R, R<sub>1</sub> and R<sub>2</sub> are other than hydrogen and no more than two of R, R<sub>1</sub> and R<sub>2</sub> are other than hydrogen on any one ring, and

ii) R<sub>1</sub> and R<sub>2</sub> on ring A are not both halo, and

iii) no two trifluoromethyl groups are on adjacent carbon atoms.

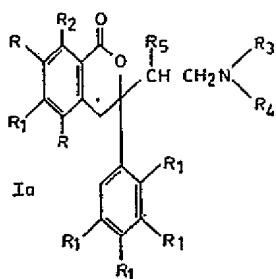
The invention also provides processes for the production of compounds of formula I, which comprise

a) cyclising by heating to at least 100°C a compound of formula II,



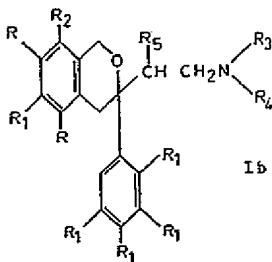
in which  
 R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> and the provisos are as defined above, and  
 R<sub>7</sub> signifies alkyl of 1 to 5 carbon atoms, alkenyl of 2 to 5 carbon atoms or benzyl, to form a compound of formula Ia,

[Price 25p]



in which  
 $R$ ,  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$  and  $R_5$  and the provisos  
 are as defined above, or

- 5 b) reducing using an alkali metal borohydride, in an inert organic solvent and at a temperature of  $-20$  to  $80^\circ\text{C}$  and in the presence of boron trifluoride etherate a compound of formula Ia as defined above, and  
 10 10 treating the resulting adduct with concentrated acid at a temperature of from  $40^\circ\text{C}$  to the reflux temperature of the reaction mixture, to form a compound of formula Ib,



- 15 in which  
 $R$ ,  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$  and  $R_5$  and the provisos  
 are as defined above.

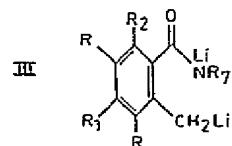
Process variant a) is preferably carried out in an inert organic solvent such as an ether, e.g. ethyl ether, or a hydrocarbon or halogenated hydrocarbon, e.g. hexane, heptane, benzene, toluene or *o*-dichlorobenzene. The preferred reaction temperature is from  $140$  to  $160^\circ\text{C}$ . Although temperatures as high as  $220$  or  $250^\circ\text{C}$  may be used. Reaction times are usually about  $15$  to  $48$  hours, under preferred conditions about  $20$  to  $28$  hours. Carrying the reaction out in the absence of oxygen, for example under inert atmosphere, such as under nitrogen, tends to increase yields and give a better quality product.

In process variant b) the compound of formula Ia may be in free base or acid addition salt form. The reducing agent is preferably sodium or lithium borohydride. The inert organic solvent is suitably tetrahydrofuran or diethyleneglycoldimethylether (diglyme). The reaction is preferably carried out at elevated temperature, especially from  $50$  to  $60^\circ\text{C}$ , and the reaction may be effected for

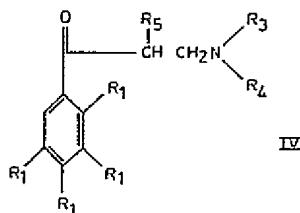
about  $1/2$  to  $2$  hours, and under preferred conditions about  $1$  hour. The particular solvent or temperature used is not critical. In the treatment of the adduct with concentrated acid, the acid is suitably concentrated hydrochloric or sulphuric acid, or, preferably, glacial acetic acid. The reaction may be carried out in an inert organic solvent such as ether or tetrahydrofuran, and is preferably carried out at the reflux temperature of the reaction mixture. Neither the temperature nor the solvents used are critical. In this process, it is convenient to remove the solvent used in the first step, without isolation of the intermediate, and introduce suitable solvent for the second step.

The compounds of formula Ia and Ib may be recovered and purified using conventional techniques such as crystallization.

The compounds of formula II may be prepared by condensing a compound of formula III,



in which  
 $R$ ,  $R_1$ ,  $R_2$  and  $R_7$  are as defined above,  
 65 with a compound of formula IV,



in which  
 $R_1$ ,  $R_5$ ,  $R_4$  and  $R_7$  are as defined above,  
 70 in an inert organic solvent and in the absence of oxygen, and hydrolyzing the reaction product:

The inert solvent is suitably diethyl ether, tetrahydrofuran, hexane, heptane or benzene, and the reaction is conveniently carried out under an inert gas, e.g. nitrogen. Suitable reaction temperatures are from  $-80$  to  $-20^\circ\text{C}$ , preferably  $-60$  to  $-40^\circ\text{C}$ , and it is preferred to add the compound of formula IV in inert solvent to a cold ( $-60$  to  $-40^\circ\text{C}$ ) inert solvent solution of the compound of formula III. Reaction times are about  $1$  to  $3$  hours. The hydrolysis is effected in conventional manner, preferably with aqueous ammonium chloride and preferably at a temperature of about  $-20$  to  $0^\circ\text{C}$ . The solvents, temperatures and hydrolysing agents used are not critical. The compounds of

45

50

55

60

65

70

75

80

85

formula II may be recovered and purified using conventional techniques.

Certain of the compounds of formulae III and IV are known and may be prepared by methods disclosed in the literature. Those compounds not specifically disclosed may be prepared by analogous methods from known materials.

It will be understood that certain of the compounds of formulae I and II exist in racemic form, or in the form of optically active isomers. Additionally, some of the compounds of formula I, particularly those in which R<sub>5</sub> signifies alkyl, may also exist as diastereomeric isomers. The separation and recovery of the isomers may be accomplished using conventional techniques and such isomers are included within the scope of this invention.

The compounds of formula I possess pharmacological activity. More particularly, they possess diuretic activity as indicated, e.g., by their activity in the unanesthetized rat when tested basically as described by R. Aston, *Toxicol. and Appl. Pharmacol. I*, 277 (1959).

The compounds also possess hypotensive/antihypertensive activity as indicated, e.g., by their activity in hypertensive rat when tested basically as described by A. Grollman (*Proc. Soc. Exptl. Biol. and Med. 57*; 103 (1944)).

The compounds are therefore indicated for use as hypotensive/antihypertensives or diuretics. The indicated daily dosage for both indicated uses is in the range from 50 to 1500 mg, conveniently administered in divided or unit doses containing from 12.5 to 750 mg of active compound, 2 to 4 times a day, or in sustained release form.

A particularly interesting compound is 3 - [2 - (dimethylamino)ethyl] - 3 - phenyl isochroman.

For the above indicated uses, the compounds of formula I may be administered in pharmaceutically acceptable acid addition salt form. Such salt forms possess the same order of activity as the free base, and are readily prepared by reacting the free base with an appropriate acid and accordingly are included within the scope of the invention. Suitable such salt forms include mineral acid salts such as the hydrochloride, hydrobromide, sulphate and phosphate and organic acid salts such as the succinate, benzoate, acetate, *p* - toluenesulphonate and benzenesulphonate. Acid addition salt forms may be converted into free base form by conventional methods such as treatment of an aqueous solution with a base such as sodium hydroxide.

The invention also provides a pharmaceutical composition comprising a compound of formula I, in free base or pharmaceutically acceptable acid addition salt form, in association with a pharmaceutically acceptable carrier or diluent.

A representative formulation suitable for oral administration is a capsule prepared by standard techniques which contains the following:

Ingredient	Parts by Weight	70
Compound of formula I, e.g.		
3 - [2 - (dimethylamino)ethyl] -		
3,4 - dihydro - 3 - phenyl		
isocoumarin	25	75
Inert filler (starch, kaolin, lactose		
etc.)	275	
		80

The following Examples 2 and 3 illustrate the invention. Example 1 illustrates the production of intermediates.

#### EXAMPLE 1

*o* - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - methyl benzamide (compound of formula II)

To a flask equipped with a stirrer, dropping funnel, condenser and gas inlet tube and maintained under a nitrogen atmosphere there is added at room temperature 40.0 g (0.28 mole) of *o* - methyl - N - methyl benzamide and 250 ml of anhydrous tetrahydrofuran. The reaction flask is immersed in an ice bath and cooled to an internal temperature of 5°C. Stirring is initiated and 380 ml of 1.6 molar *n* - butyllithium (0.616 mole) in hexane is added dropwise in *ca*. 1 hour, maintaining the temperature below 8°C. The resulting red dilithio salt is stirred at 5°C for 1 additional hour and the reaction flask is then immersed in a dry-ice/acetone bath and cooled to an internal temperature of -60°C. To the cold reaction mixture a solution of 49.7 g (0.28 mole) 3 - dimethylaminopropiophenone in 140 ml anhydrous tetrahydrofuran is added dropwise in *ca*. 45 minutes maintaining the temperature between -60 and -50°C. The resulting reaction mixture is stirred at -60°C for 1 hour, allowed to warm to 0°C in *ca*. 1 hour, and then treated with 200 ml of saturated aqueous ammonium chloride while maintaining the temperature below 10°C. The resulting solid is filtered, washed thoroughly with water and recrystallised from methylene chloride/ether (1:1) to give *o* - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - methyl benzamide; m.p. 139.5 - 140.5°C.

When the above process is carried out and

a) *o* - methyl - N - allyl benzamide, or  
b) *o* - methyl - N - benzyl benzamide is used in place of *o* - methyl - N - methyl benzamide, there is obtained

a) *o* - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - allyl benzamide, or

b) *o* - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - benzyl benzamide, respectively.

When the above detailed process is carried out and in place of o - methyl - N - methyl benzamide there is used

- 5      c) 2 - methyl - 6 - methoxy - N - methyl benzamide,  
       d) 4 - chloro - 2 - methyl - N - methyl benzamide,  
       e) 2,3 - dimethyl - N - methyl benzamide, or

10     f) 2 - methyl - 5 - trifluoromethyl - N - methyl benzamide, there is obtained  
       g) 2 - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - 6 - methoxy - N - methyl benzamide,

15     h) 4 - chloro - 2 - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - methyl benzamide,

20     i) 2 - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - 3,N - dimethyl benzamide, or  
       f) 2 - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - 5 - trifluoromethyl - N - methyl benzamide, respectively.

25     When the above detailed procedure is carried out and in place of 3 - dimethylamino-propiophenone there is used

30     g) 3',4' - dichloro - 3 - dimethylamino-propiophenone,  
       h) 3 - dimethylamino - 4' - methoxy-propiophenone,

35     i) 3 - (N - methylpiperazino)propiophenone,  
       j) 3 - morpholinopropiophenone,  
       k) 3 - dimethylamino - 2 - methylpropiophenone,

40     l) 3 - thiomorpholinopropiophenone,  
       m) 3 - pyrrolidylpropiophenone,  
       n) 3 - piperidinylpropiophenone,  
       o) 3 - diallylamo - 2' - methylpropiophenone,

45     p) 3 - dibenzylamino - 3' - trifluoromethyl-propiophenone, or  
       q) 3 - dimethylamino - 3',4' - methylenedioxypropiophenone, there is obtained

50     g) o - {3,4 - dichloro -  $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - methyl benzamide, m.p. 130—131°C,  
       h) o - { $\alpha$  - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxy - p - methoxy}phenethyl - N - methyl benzamide,

55     i) o - { $\alpha$  - hydroxy -  $\alpha$  - [2 - (N - methylpiperazino)ethyl]}phenethyl - N - methyl benzamide,  
       j) o - { $\alpha$  - hydroxy -  $\alpha$  - [2 - (morpholino)ethyl]}phenethyl - N - methyl benzamide,

60     k) o - { $\alpha$  - (2 - dimethylamino - 1 - methyl)ethyl] -  $\alpha$  - hydroxy}phenethyl - N - methyl benzamide,  
       l) o - { $\alpha$  - hydroxy -  $\alpha$  - [2 - (thiomorpholino)ethyl]}phenethyl - N - methyl benzamide,  
       m) o - { $\alpha$  - hydroxy -  $\alpha$  - [2 - (pyrrolidyl)ethyl]}phenethyl - N - methyl benzamide,  
       n) o - { $\alpha$  - hydroxy -  $\alpha$  - [2 - (piperidinyl)ethyl]}phenethyl - N - methyl benzamide,

- o)  $\text{o} - \{\alpha - [2 - (\text{diallylamino})\text{ethyl}] -$  65  
 $\alpha - \text{hydroxy} - \text{o} - \text{methyl}\}$  phenethyl - N -  
 methyl benzamide,  
 p)  $\text{o} - \{\alpha - [2 - (\text{dibenzylamino})\text{ethyl}] -$   
 $\alpha - \text{hydroxy} - \text{m} - \text{trifluoromethyl}\}$  phenethyl -  
 N - methyl benzamide, or 70  
 q)  $\text{o} - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] -$   
 $\alpha - \text{hydroxy} - 3,4 - \text{methylenedioxy}\}$  phen-  
 ethyl - N - methyl benzamide, respectively.

### EXAMPLE 2

- 3 - [2 - (Dimethylamino)ethyl] - 3,4 - 75  
 dihydro - 3 - phenyl isocoumarin [process  
 variant a)]

To a flask equipped with a stirrer, condenser and gas inlet tube and maintained under a nitrogen atmosphere there is added at room temperature 16.3 g (0.05 mole) of *o* - { $\alpha$  - [2 - (dimethylamino)ethyl]} -  $\alpha$  - hydroxyphenethyl - *N* - methyl benzamide and 170 ml of *o* - dichlorobenzene. Stirring is initiated and the mixture is heated at reflux for 18 hours. The excess *o* - dichlorobenzene is then removed by distillation in vacuo and the resulting oil is crystallized from ether to give 3 - [2 - (dimethylamino)ethyl] - 3,4 - dihydro - 3 - phenyl isocoumarin; m.p. 95.0-95.5°C.

When the above process is carried out and in place of o - [2 - (dimethylamino)ethyl] -  $\alpha$  - hydroxyphenethyl - N - methyl benzamide there is used

- a)  $\sigma - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - \text{N} - \text{allyl benzamide}$ ,  
 b)  $\sigma - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - \text{N} - \text{benzyl benzamide}$ ,  
 c)  $2 - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - 6 - \text{methoxy} - \text{N} - \text{methylbenzamide}$ ,  
 d)  $4 - \text{chloro} - 2 - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - \text{N} - \text{methylbenzamide}$ ,  
 e)  $2 - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - 3, \text{N} - \text{dimethyl benzamide}$ ,  
 f)  $2 - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - 5 - \text{trifluoromethyl} - \text{N} - \text{methyl benzamide}$ ,  
 g)  $\sigma - \{3,4 - \text{dichloro} - \alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy}\} \text{phenethyl} - \text{N} - \text{methyl benzamide}$ ,  
 h)  $\sigma - \{\alpha - [2 - (\text{dimethylamino})\text{ethyl}] - \alpha - \text{hydroxy} - p - \text{methoxy}\} \text{phenethyl} - \text{N} - \text{methyl benzamide}$ ,  
 i)  $\sigma - \{\alpha - \text{hydroxy} - \alpha - [2 - (\text{N} - \text{methylpiperazino})\text{ethyl}]\} \text{phenethyl} - \text{N} - \text{methyl benzamide}$ ,  
 j)  $\sigma - \{\alpha - \text{hydroxy} - \alpha - [2 - (\text{morpholino})\text{ethyl}]\} \text{phenethyl} - \text{N} - \text{methyl benzamide}$ ,  
 k)  $\sigma - \{\alpha - [2 - (\text{dimethylamino} - 1 - \text{methylethyl})] - \alpha - \text{hydroxy}\} \text{phenethyl} - \text{N} - \text{methyl benzamide}$ ,

- 1)  $\sigma - \{\alpha - \text{hydroxy} - \alpha - [2 - \text{thiomorpholinoethyl}] \text{phenethyl} - N - \text{methyl benzamide},$   
     m)  $\sigma - \{\alpha - \text{hydroxy} - \alpha - [2 - (\text{pyrrolidyl})\text{ethyl}] \text{phenethyl} - N - \text{methyl benzamide},$   
         n)  $\sigma - \{\alpha - \text{hydroxy} - \alpha - [2 - (\text{piperidinyl})\text{ethyl}] \text{phenethyl} - N - \text{methyl benzamide},$   
         o)  $\sigma - \{\alpha - [2 - (\text{diallylaminoethyl}) - \alpha - \text{hydroxy} - o - \text{methyl}] \text{phenethyl} - N - \text{methyl benzamide},$   
             p)  $\sigma - \{\alpha - [2 - (\text{dibenzylaminoethyl}) - \alpha - \text{hydroxy} - m - \text{trifluoromethyl}] \text{phenethyl} - N - \text{methyl benzamide, or}$   
             q)  $\sigma - \{\alpha - [2 - (\text{dimethylaminoethyl}) - \alpha - \text{hydroxy} - 3,4 - \text{methylenedioxy}] \text{phenethyl} N - \text{methyl benzamide, there is obtained}$   
                 a) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin, m.p. 95.0—95.5°C,  
                 b) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin, m.p. 95.0—95.5°C,  
                 c) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 8 - methoxy - 3 - phenyl isocoumarin,  
                 d) 6 - chloro - 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin,  
                 e) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 5 - methyl - 3 - phenyl isocoumarin,  
                 f) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl - 7 - trifluoromethyl isocoumarin,  
                 g) 3 - (3,4 - dichlorophenyl) - 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro isocoumarin, m.p. in hydrochloride salt form, 286—287°C,  
                 h) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - (p - methoxyphenyl) isocoumarin, m.p. 87—88°C,  
                 i) 3,4 - dihydro - 3 - [2 - (N - methylpiperazinoethyl) - 3 - phenyl isocoumarin, m.p. 134—135.5°C,  
                 j) 3,4 - dihydro - 3 - [2 - (\text{morpholinoethyl}) - 3 - phenyl isocoumarin, m.p. 120—121°C,  
                 k) 3 - {[2 - (\text{dimethylamino}) - 1 - methyl] - ethyl} - 3,4 - dihydro - 3 - phenyl isocoumarin, m.p. 132—134°C,  
                 l) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{thiomorpholinoethyl})] isocoumarin,  
                 m) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{pyrrolidylethyl})] isocoumarin,  
                 n) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{piperidinylethyl})] isocoumarin,  
                 o) 3 - [2 - (\text{diallylaminoethyl}) - 3,4 - dihydro - 3 - (o - methylphenyl)isocoumarin, m.p. 173.5—174.5°C,  
                 p) 3 - [2 - (\text{dibenzylaminoethyl}) - 3,4 - dihydro - 3 - (m - trifluoromethylphenyl)isocoumarin, or  
                 q) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - (3,4 - methylenedioxyphenyl)isocoumarin, respectively.

## EXAMPLE 3

65

3 - [2 - (\text{Dimethylaminoethyl}) - 3 - phenyl isochroman (process variant b))

To a solution of 13.0 g (0.042 mole) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin in 220 ml diglyme was added in one portion 177 g (160 ml) (0.126 mole) of boron trifluoride etherate.

The resulting mixture was added dropwise with stirring to a solution of 3.2 g sodium borohydride (0.084 mole) in 220 ml diglyme, maintaining the temperature at 0°C. After the addition was complete the resulting mixture was heated at 55°C for 1 hour and then cooled in ice and treated dropwise with 100 ml water, maintaining the temperature at about 5°C. The solvents were removed *in vacuo* and the residue treated with ether. The insoluble boronhydride adduct was dissolved in 320 ml tetrahydrofuran containing 120 ml glacial acetic acid and refluxed for 4 hours.

The solvents were removed *in vacuo* and the residue dissolved in water and made basic by the addition of solid potassium hydroxide and extracted with ether. The ether was dried over anhydrous magnesium sulfate and filtered, cooled in ice and treated with gaseous hydrogen chloride and the resulting solid was filtered and recrystallized from methylene chloride/ether to give the product 3 - [2 - (\text{dimethylaminoethyl}) - 3 - phenyl isochroman hydrochloride, m.p. 164.5—165.0°C.

When the above process is carried out and in place of 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin there is used

a) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 8 - methoxy - 3 - phenyl isocoumarin,

b) 6 - chloro - 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl isocoumarin,

c) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 5 - methyl - 3 - phenyl isocoumarin,

d) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - phenyl - 7 - trifluoromethyl isocoumarin,

e) 3 - (3,4 - dichlorophenyl) - 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro isocoumarin,

f) 3 - [2 - (\text{dimethylaminoethyl}) - 3,4 - dihydro - 3 - (p - methoxyphenyl)isocoumarin,

g) 3,4 - dihydro - 3 - (2 - (N - methylpiperazinoethyl) - 3 - phenyl isocoumarin,

h) 3,4 - dihydro - 3 - [2 - (\text{morpholinoethyl}) - 3 - phenyl isocoumarin,

i) 3 - {[2 - (\text{dimethylamino}) - 1 - methyl] - ethyl} - 3,4 - dihydro - 3 - phenyl isocoumarin,

j) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{thiomorpholinoethyl})] isocoumarin,

k) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{pyrrolidylethyl})] isocoumarin,

l) 3,4 - dihydro - 3 - phenyl - 3 - [2 - (\text{piperidinylethyl})] isocoumarin,

70

80

85

90

95

100

105

110

115

120

125

130

135

140

145

150

155

160

165

170

175

180

185

190

195

200

205

210

215

220

225

230

235

240

245

250

255

260

265

270

275

280

285

290

295

300

305

310

315

320

325

330

335

340

345

350

355

360

365

370

375

380

385

390

395

400

405

410

415

420

425

430

435

440

445

450

455

460

465

470

475

480

485

490

495

500

505

510

515

520

525

530

535

540

545

550

555

560

565

570

575

580

585

590

595

600

605

610

615

620

625

630

635

640

645

650

655

660

665

670

675

680

685

690

695

700

705

710

715

720

725

730

735

740

745

750

755

760

765

770

775

780

785

790

795

800

805

810

815

820

825

830

835

840

845

850

855

860

865

870

875

880

885

890

895

900

905

910

915

920

925

930

935

940

945

950

955

960

965

970

975

980

985

990

995

1000

1005

1010

1015

1020

1025

1030

1035

1040

1045

1050

1055

1060

1065

1070

1075

1080

1085

1090

1095

1100

1105

1110

1115

1120

1125

1130

1135

1140

1145

1150

1155

1160

1165

1170

1175

1180

1185

1190

1195

1200

1205

1210

1215

1220

1225

1230

1235

1240

1245

1250

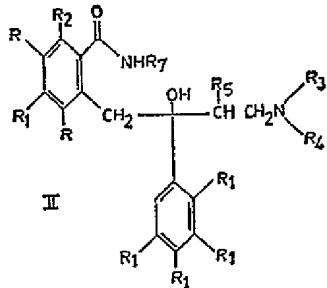
1255

1260

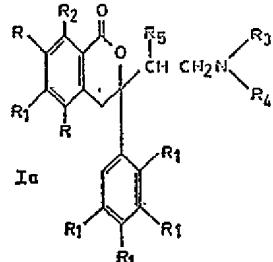
1265

- m) 3 - [2 - (diallylamino)ethyl] - 3,4 - dihydro - 3 - (*o* - methylphenyl)isocoumarin,  
 n) 3 - [2 - (dibenzylamino)ethyl] - 3,4 - dihydro - 3 - (*m* - trifluoromethylphenyl)-  
 5 isocoumarin, or  
 o) 3 - [2 - (dimethylamino)ethyl] - 3,4 - dihydro - 3 - (3,4 - methylenedioxyphenyl)-  
 isocoumarin, respectively there is obtained as  
 the hydrochloride  
 10 a) 3 - [2 - (dimethylamino)ethyl] - 8 - methoxy - 3 - phenyl isochroman,  
 b) 6 - chloro - 3 - [2 - (dimethylamino)ethyl] - 3 - phenyl isochroman,  
 c) 3 - [2 - (dimethylamino)ethyl] - 5 - methyl - 3 - phenyl isochroman,  
 d) 3 - [2 - (dimethylamino)ethyl] - 3 - phenyl - 7 - trifluoromethyl isochroman,  
 e) 3 - (3,4 - dichlorophenyl) - 3 - [2 - (dimethylamino)ethyl] isochroman,  
 20 f) 3 - [2 - (dimethylamino)ethyl] - 3 - (*p* - methoxyphenyl)isochroman, m.p. in succinate salt form 136.5-137.5°C,  
 g) 3 - [2 - (*N* - methyldiethylamino)ethyl] - 3 - phenyl isochroman,  
 25 h) 3 - [2 - (morpholino)ethyl] - 3 - phenyl isochroman,  
 i) 3 - {[2 - (dimethylamino) - 1 - methyl]ethyl} - 3 - phenyl isochroman,  
 j) 3 - phenyl - 3 - [2 - (thiomorpholino)ethyl]isochroman,  
 30 k) 3 - phenyl - 3 - [2 - (pyrrolidyl)ethyl]-isochroman,  
 l) 3 - phenyl - 3 - [2 - (piperidinyl)ethyl]-isochroman,  
 35 m) 3 - [2 - (diallylamino)ethyl] - 3 - (*o* - methylphenyl)isochroman,  
 n) 3 - [2 - (dibenzylamino)ethyl] - 3 - (*m* - trifluoromethylphenyl)isochroman, or  
 o) 3 - [2 - (dimethylamino)ethyl] - 3 - (3,4 - methylenedioxyphenyl)isochroman, res-  
 pectively.

R<sub>1</sub>'s on adjacent carbon atoms together signify methylenedioxy, 55  
 R<sub>2</sub> signifies hydrogen, trifluoromethyl, alkoxy of 1 to 5 carbon atoms, fluorine or chlorine, 60  
 R<sub>3</sub> and R<sub>4</sub> independently signify alkyl of 1 to 5 carbon atoms, alkenyl of 2 to 5 carbon atoms or benzyl, or  
 R<sub>3</sub> and R<sub>4</sub> together signify a -(CH<sub>2</sub>)— chain of 4 to 7 carbon atoms or  
 —(CH<sub>2</sub>)<sub>2</sub>—Z—(CH<sub>2</sub>)<sub>2</sub>—  
 in which Z signifies oxygen or sulphur or nitrogen substituted by alkyl of 1 to 5 carbon atoms, 65  
 R<sub>5</sub> signifies hydrogen or straight chain alkyl of 1 to 5 carbon atoms and  
 X signifies —CH<sub>2</sub>— or —CO—, with the provisos that  
 i) no more than three of R<sub>1</sub>, R<sub>2</sub> and R<sub>5</sub> are other than hydrogen and no more than two of R<sub>3</sub>, R<sub>4</sub> and R<sub>2</sub> are other than hydrogen on any one ring, and  
 ii) R<sub>1</sub> and R<sub>2</sub> on ring A are not both halo, and  
 iii) no two trifluoromethyl groups are on adjacent carbon atoms, which comprise  
 a) cyclising by heating to at least 100°C a compound of formula II, 70  
 80



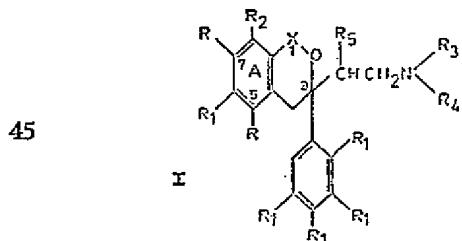
in which  
 R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> and the provisos are as defined above, and  
 R<sub>7</sub> signifies alkyl of 1 to 5 carbon atoms, 85  
 alkenyl of 2 to 5 carbon atoms or benzyl, to form a compound of formula Ia,



in which  
 R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> and the provisos are as defined above, or 90

#### WHAT WE CLAIM IS:—

1. Processes for the production of compounds of formula I,

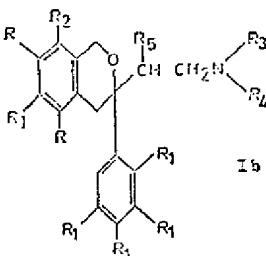


- in which each R independently signifies hydrogen, trifluoromethyl or alkyl or alkoxy of 1 to 5 carbon atoms, each  
 45 50 R<sub>1</sub> independently signifies hydrogen, fluorine, chlorine, trifluoromethyl or alkyl or alkoxy of 1 to 5 carbon atoms, or two

in which  
 R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> and the provisos are as defined above, or

- b) reducing using an alkali metal borohydride, in an inert organic solvent and at a temperature of -20 to 80°C and in the presence of boron trifluoride etherate a compound of formula Ia as defined above, and treating the resulting adduct with concentrated acid at a temperature of from 40°C to the reflux temperature of the reaction mixture, to form a compound of formula Ib,

10



Ib

in which

R, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> and the provisos are as defined above.

- 15 2. A process according to claim 1, in which the compound of formula II is heated to a temperature of from 140° to 160°C, in an inert organic solvent and under an inert atmosphere.

- 20 3. A process according to claim 1, in which the compound of formula Ia is reduced with sodium or lithium borohydride and the resulting adduct is treated with concentrated hydrochloric or sulphuric acid or with glacial acetic acid.

- 25 4. A process according to Claim 1, 2 or 3, in which a resulting free base form of the compound of formula I is converted into an acid addition salt form, or *vice versa*.

- 30 5. A process according to Claim 1, substantially as hereinbefore described with reference to Example 2 or 3.

6. A compound of formula I, whenever prepared by a process according to any one of Claims 1 to 5.

- 35 7. A compound of formula I, as defined in Claim 1.

8. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 3 - phenyl isocoumarin.

- 40 9. 3 - (3,4 - Dichlorophenyl) - 3 - [2 - (dimethylamino)ethyl] - 3,4 - dihydro isocoumarin.

10. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 3 - (p - methoxyphenyl)isocoumarin.

- 45 11. 3,4 - Dihydro - 3 - [2 - (N - methylpiperazino)ethyl] - 3 - phenyl isocoumarin.

12. 3,4 - Dihydro - 3 - [2 - (morpholino)ethyl] - 3 - phenyl isocoumarin.

- 50 13. 3 - {[2 - (Dimethylamino) - 1 - methyl]ethyl} - 3,4 - dihydro - 3 - phenyl isocoumarin.

14. 3 - [2 - (Dimethylamino)ethyl] - 3 - phenyl isochroman.

15. 3 - [2 - (Dimethylamino)ethyl] - 3 - (p - methoxyphenyl)isochroman.

16. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 8 - methoxy - 3 - phenyl isocoumarin.

17. 6 - Chloro - 3 - [2 - (dimethylamino)ethyl] - 3,4 - dihydro - 3 - phenyl isocoumarin.

18. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 5 - methyl - 3 - phenyl isocoumarin.

19. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 3 - phenyl - 7 - trifluoromethyl isocoumarin.

20. 3,4 - Dihydro - 3 - phenyl - 3 - [2 - (thiomorpholino)ethyl]isocoumarin.

21. 3,4 - Dihydro - 3 - phenyl - 3 - [2 - (pyrrolidyl)ethyl]isocoumarin.

22. 3,4 - Dihydro - 3 - phenyl - 3 - [2 - (piperidyl)ethyl]isocoumarin.

23. 3 - [2 - (Dibenzylamino)ethyl] - 3,4 - dihydro - 3 - (m - trifluoromethylphenyl)isocoumarin.

24. 3 - [2 - (Dimethylamino)ethyl] - 3,4 - dihydro - 3 - (3,4 - methylenedioxypyphenyl)isocoumarin.

25. 3 - [2 - (Dimethylamino)ethyl] - 8 - methoxy - 3 - phenyl isochroman.

26. 6 - Chloro - 3 - [2 - (dimethylamino)ethyl] - 3 - phenyl isochroman.

27. 3 - [2 - (Dimethylamino)ethyl] - 5 - methyl - 3 - phenyl isochroman.

28. 3 - [2 - (Dimethylamino)ethyl] - 3 - phenyl - 7 - trifluoromethyl isochroman.

29. 3 - (3,4 - Dichlorophenyl) - 3 - [2 - (dimethylamino)ethyl]isochroman.

30. 3 - [2 - (N - methylpiperazino)ethyl] - 3 - phenyl isochroman.

31. 3 - [2 - (Morpholino)ethyl] - 3 - phenyl isochroman.

32. 3 - {[2 - (dimethylamino) - 1 - methyl]ethyl} - 3 - phenyl isochroman.

33. 3 - Phenyl - 3 - [2 - (thiomorpholino)ethyl]isochroman.

34. 3 - Phenyl - 3 - [2 - (pyrrolidyl)ethyl]isochroman.

35. 3 - Phenyl - 3 - [2 - (piperidinyl)ethyl]isochroman.

36. 3 - [2 - (Diallylarnino)ethyl] - 3 - (o - methylphenyl)isochroman.

37. 3 - [2 - (Dibenzylamino)ethyl] - 3 - (m - trifluoromethylphenyl)isochroman.

38. 3 - [2 - (Dimethylamino)ethyl] - 3 - (3,4 - methylenedioxypyphenyl)isochroman.

39. 3 - [2 - (Diallylarnino)ethyl] - 3,4 - dihydro - 3 - (o - methylphenyl)isochroman.

40. A compound according to any one of Claims 6 to 39, in acid addition salt form.

110

41. A pharmaceutical composition comprising a compound according to any one of Claims 6 to 39, in free base or pharmaceutically acceptable acid addition salt form, in association with a pharmaceutically acceptable carrier or diluent.

B. A. YORKE & CO.,  
Chartered Patent Agents,  
98, The Centre,  
Feltham,  
Middlesex, TW13 4EP.  
Agents for the Applicants.

Printed for Her Majesty's Stationery Office, by the Courier Press, Leamington Spa, 1974.  
Published by The Patent Office, 25 Southampton Buildings, London, WC2A 1AY, from  
which copies may be obtained.